

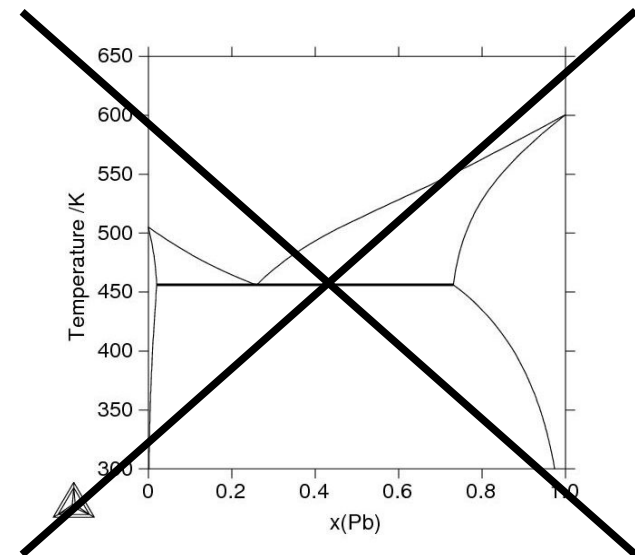


Thermodynamic databases for lead free solders

**Alan Dinsdale
Materials Division, NPL, UK**

Outline

- Why lead-free solders?
- COST and COST Actions
- Database construction
- COST 531 & MP0602 databases for lead-free solders
 - Scope of databases
 - Modelling issues
 - Applications



So, why lead-free solders?

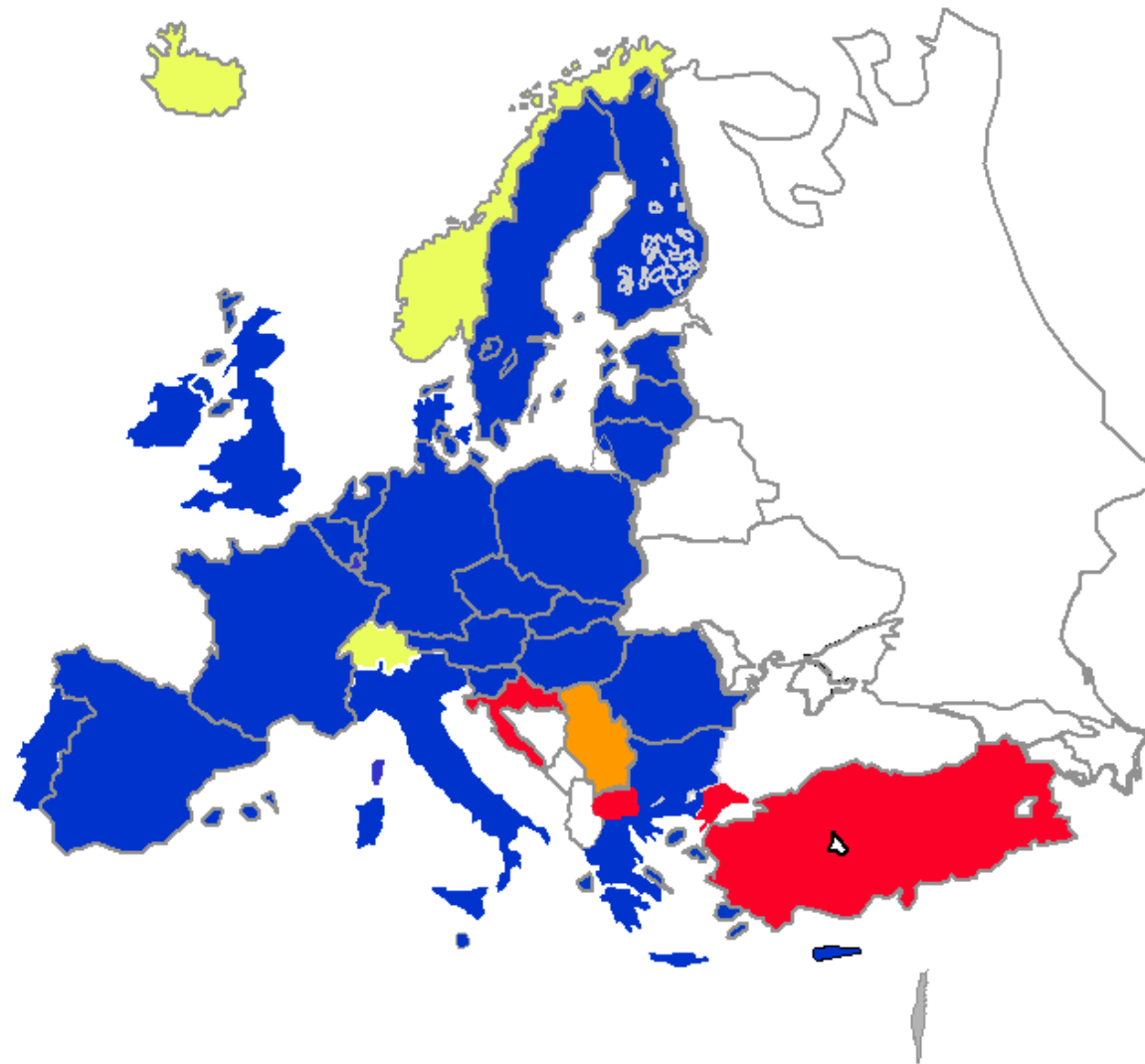
- Environmental issues
 - Toxic
 - Incorporated into bone marrow, nerve tissue, brain, kidneys..
 - Low levels affect the brain, central nervous system, lead to learning difficulties in children, impair physical and mental development
- Disposal of electrical and electronic equipment
 - Lead leaching into groundwater from landfill
- European legislation
 - Waste Electrical and Electronic Equipment (WEEE) Directive
 - Restriction of the Use of Certain Hazardous Substances in Electrical and Electronic Equipment Regulations 2006 (RoHS) – 1st July 2006
- Solders
 - Conventional “electrician’s solder” already replaced
 - Need replacement for high temperature, high-Pb containing alloys
 - Currently exempt from RoHS legislation but future EU legislation is likely

COST and COST Actions

What is COST ?

- European **CO**operation in the field of **S**cientific and **T**echnical research
- **Mission**: To strengthen Europe in scientific and technical research for peaceful purposes through the support of cooperation and interaction between European researchers
- **34** COST member states plus **1** cooperating member
- Supported by the European Commission through the **Framework Programme**
- Works through a series of **actions** in the form of networked research projects
- Aims to coordinate nationally funded research at a European level

COST Countries



◆ 27 EU Member States

◆ Candidate countries

- ▶ Croatia*
- ▶ Former Yugoslav Republic of Macedonia (FYROM)*
- ▶ Turkey

◆ EFTA Member States

- ▶ Iceland
- ▶ Norway
- ▶ Switzerland

◆ Other European Countries

- ▶ Republic of Serbia*

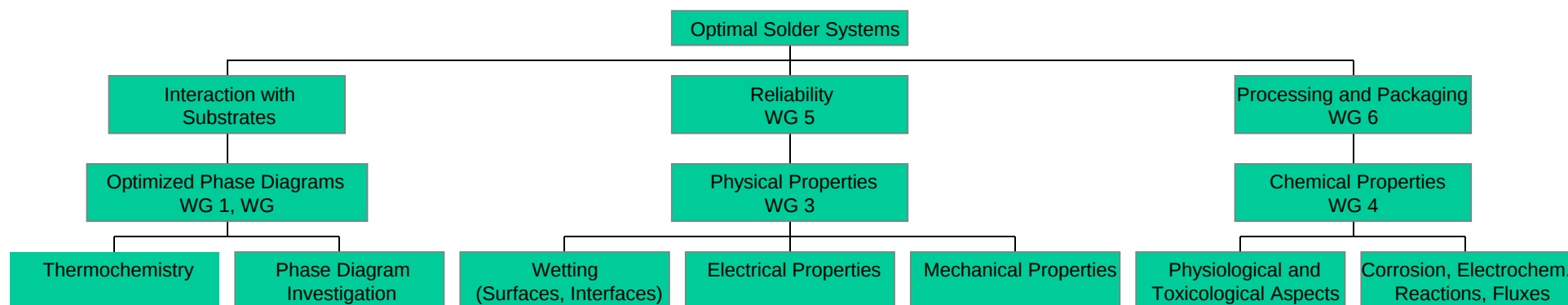
◆ COST Cooperating States

- ▶ Israel

- A European Action on Lead-free Soldering (2002 to 2007)
- Coordinated by Herbert Ipser (University of Vienna)
- 45 research institutes from 17 countries
- Working Groups
 - Groups 1&2: Thermochemistry and Phase Diagrams
(Experimental + Modelling)
 - Groups 3-6: Physical, Chemical Properties,
Reliability Issues

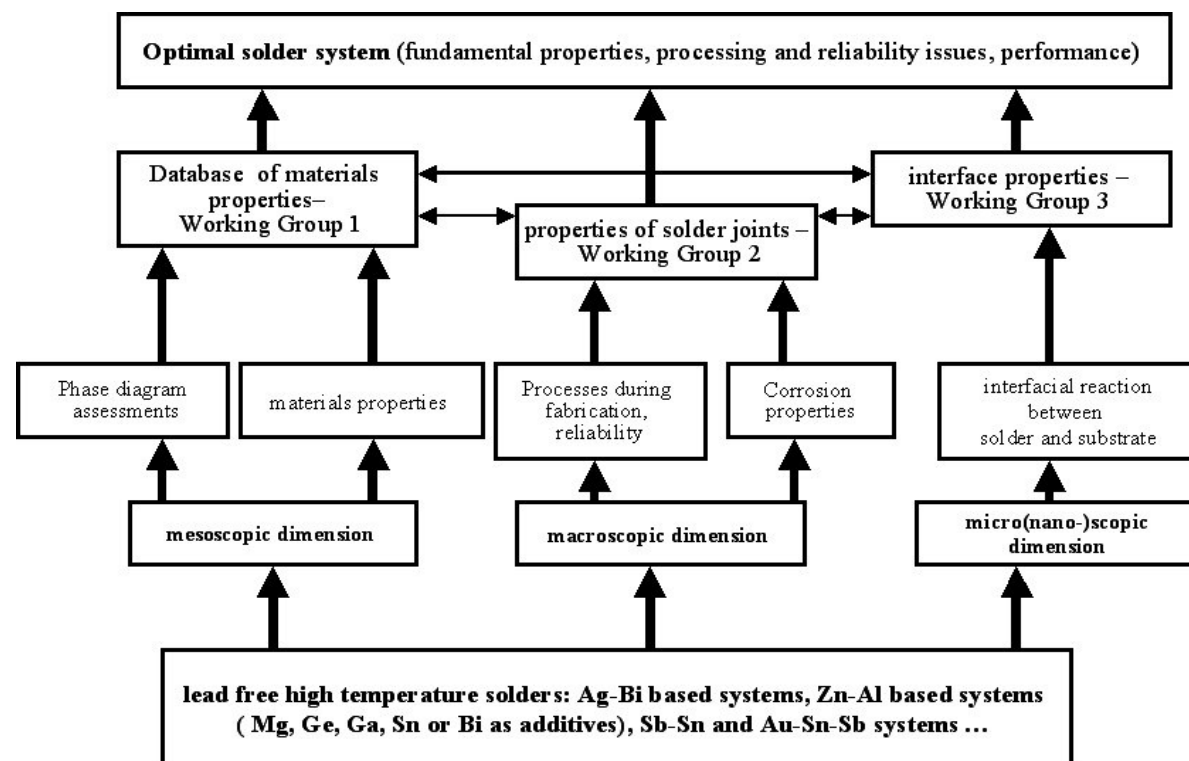
COST 531 – Lead-free Solder Materials: 2002-2007

“....to increase the basic knowledge on possible alloy systems that can be used as lead-free solder materials and to provide a scientific basis for a decision which of these materials to use for different soldering purposes in order to replace the currently used lead-containing solders in the future.”



COST MP0602 - Advanced Solder Materials for High Temperature Application (HISOLD): 2007-2011

- “...the investigation of Pb-free replacements for high-Pb solders for high-temperature applications..(using) a multiscale approach: **meso-scale:** The application of thermodynamics and kinetics; the development of materials property databases.
- **macro-scale:** The creation of a phenomenological description of corrosion and deformation processes occurring in a solder joint during fabrication and service,
- **micro- (nano-) scale:** The investigation by experiment and modelling of the initial stage of the formation of intermetallic phases at the solder/substrate interface....increase the basic understanding of alloys that can be used as Pb-free alternatives to high-temperature solders for practical applications..”



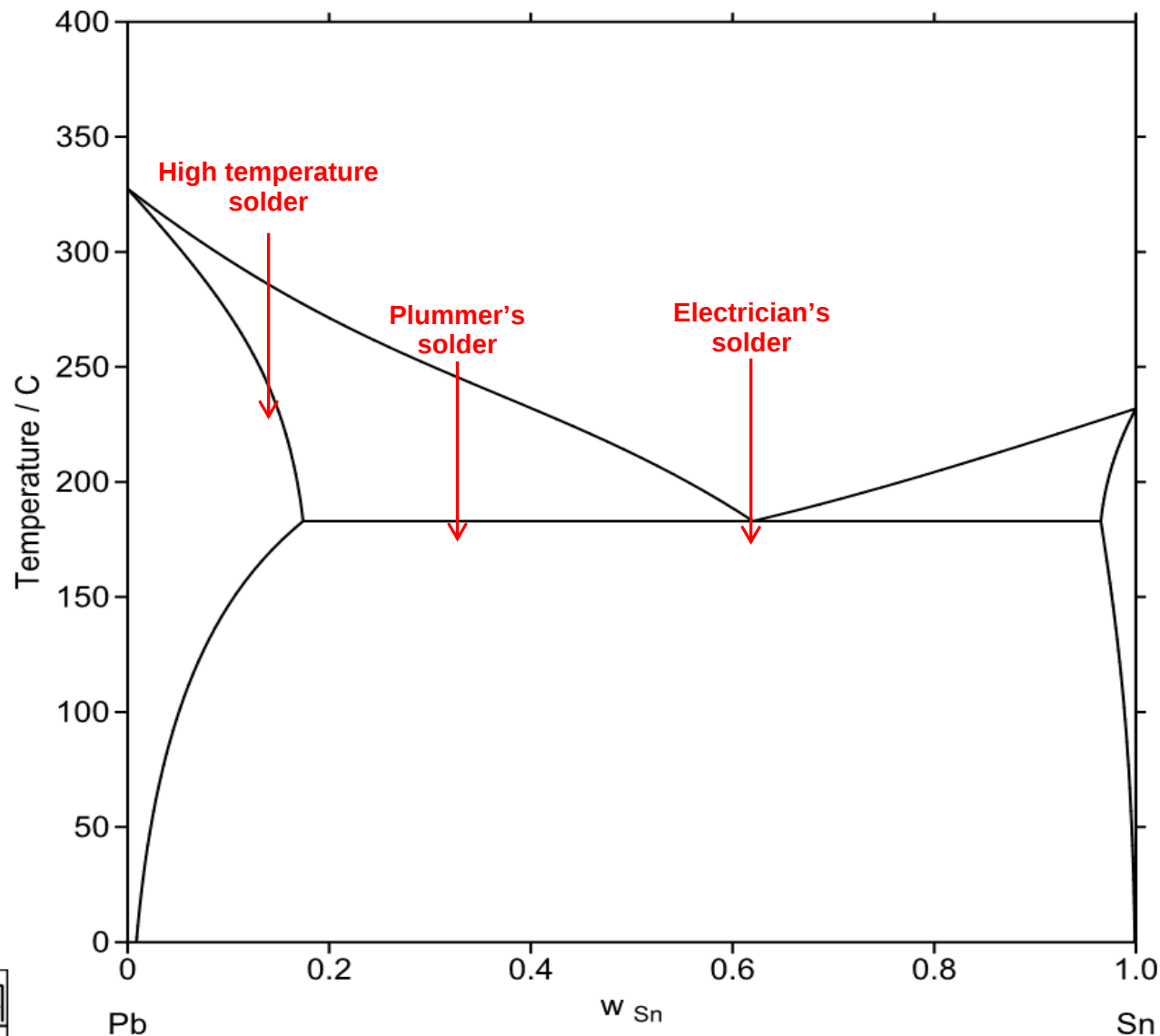
Coordinated by Ales Kroupa, Institute of Physics of Materials, Brno, Czech Republic

COST MP0602 - Advanced Solder Materials for High Temperature Application (HISOLD): 2007-2011

- Signatories

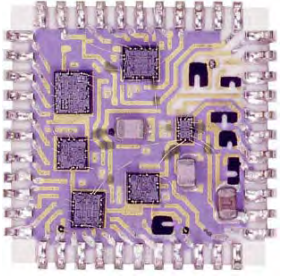
- Austria
- (Argentina)
- Belgium
- Bulgaria
- Czech Republic*
- Denmark
- France
- Finland
- Germany
- Hungary
- Ireland
- Italy
- Netherlands
- Poland
- Portugal
- Romania
- (Russian Federation)
- Serbia
- Slovak Republic
- Slovenia
- Sweden
- Switzerland
- Turkey
- (Ukraine)
- United Kingdom

Pb-Sn phase diagram



NPL
National Physical Laboratory

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Need for high temperature solders

- **Power circuits**
very high levels of conductivity required
 - **Automotive industry under bonnet applications**
high current and low voltage, high temperatures within the engine area
 - **Multi-chip modelling (MCM)**
step soldering approach - soldering of various levels of the package with different solders of different melting points
upper limit around 350°C (limit set by polymer used in substrate)
- Pb-free high temperature solders required to replace current alloys for wide range of melting temperatures**

Database construction

Database Construction: Sources of data

- SGTE solution database (SSOL)
 - *(Data sharing arrangement)*
- NPL Solders database
- Brno Solders database
 - *(Institute of Physics of Materials, Czech Academy of Sciences, Brno)*
- New assessments generated during the Actions
- Assessments in the open scientific literature
 - Reassessment often required
 - New experimental data generated during the Actions
 - New modelling
- Testing with MTDATA, ThermoCalc, Pandat

Database Construction: The need for internal consistency

- Unary data
 - Basis for all of the datasets within the database
 - SGTE Unary v4.4
 - Sn(fcc)
 - FUNCTION GFCCSN 100 5510 – 8.46*T + GHSERSN; 3000
 - Sn(hcp)
 - FUNCTION GHPCSN 100 3900 – 7.646*T + GHSERSN; 3000
 - SGTE Unary v1.1
 - Sn(fcc)
 - FUNCTION GFCCSN 100 4150 – 5.2*T + GHSERSN; 3000
 - Sn(hcp)
 - FUNCTION GHPCSN 100 3900 – 4.4*T + GHSERSN; 3000

Database Construction: The need for internal consistency

- Phase names
 - The same phase in each system *should* have the same name
 - FCC_A1, BCC_A2, HCP_A3, BCT_A5
 - CUIN_GAMMA
 - Cu-In
 - Ag-In
 - 'Common names'

Database Construction: The need for internal consistency

Phase Name	Common Name	Strukturbericht designation	Pearson Symbol
LIQUID	Liquid		
FCC_A1	(Ag)	A1	<i>cF4</i>
BCC_A2	β	A2	<i>cP2</i>
HCP_A3	ζ	A3	<i>hP2</i>
CUIN_GAMMA	γ	<i>D8₃</i>	<i>cP52</i>
AGIN2	AgIn ₂	C16	<i>tI12</i>
TETRAG_A6	(In)	A6	<i>tI2</i>

Scope of databases

Scope: 11 elements

Ag, Au, Bi, Cu, In, Ni, Pb, Pd, Sb, Sn, Zn

Binary systems: All 55 binary systems except

Ni-Sb (now available) and Pd-Sb

Selected ternary systems:

Ag-Au-Bi, Ag-Au-Sb, Ag-Bi-Sn, Ag-Cu-Ni, Ag-Cu-Pb,

Ag-Cu-Sn, Ag-In-Sn, Ag-Ni-Sn

Au-Bi-Sb, Au-In-Sb, Au-In-Sn, Au-Ni-Sn

Bi-In-Sn, Bi-Sb-Sn, Bi-Sn-Zn

Cu-In-Sn, Cu-Ni-Pb, Cu-Ni-Sn

In-Sb-Sn, In-Sn-Zn

COST 531 database - Systems

- Assessed binary systems

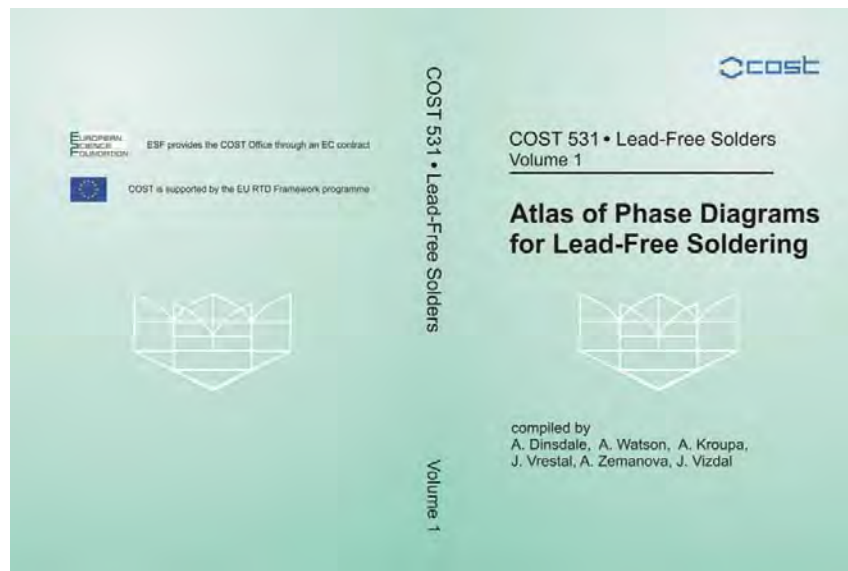
Ag-Au, Ag-Bi, Ag-Cu, Ag-In, Ag-Ni,
Ag-Pb, Ag-Pd, Ag-Sb, Ag-Sn,
Ag-Zn, Au-Bi, Au-Cu, Au-In, Au-Ni,
Au-Pb, Au-Pd, Au-Sb, Au-Sn,
Au-Zn, Bi-Cu, Bi-In, Bi-Ni, Bi-Pb,
Bi-Pd, Bi-Sb, Bi-Sn, Bi-Zn, Cu-In,
Cu-Ni, Cu-Pb, Cu-Pd, Cu-Sb,
Cu-Sn, Cu-Zn, In-Ni, In-Pb, In-Pd,
In-Sb, In-Sn, In-Zn, Ni-Pb, Ni-Pd,
Ni-Sn, Ni-Zn, Pb-Pd, Pb-Sb,
Pb-Sn, Pb-Zn, Pd-Sn, Pd-Zn,
Sb-Sn, Sb-Zn, Sn-Zn

- Assessed ternary systems

Ag-Au-Bi, Ag-Au-Sb, Ag-Bi-Sn,
Ag-Cu-In*, Ag-Cu-Ni, Ag-Cu-Pb,
Ag-Cu-Sn, Ag-In-Sn, Ag-Ni-Sn,
Au-Bi-Sb, Au-In-Sb, Au-Ni-Sn,
Bi-In-Sn, Bi-Sb-Sn, Bi-Sn-Zn,
Cu-In-Sn, Cu-Ni-Sn, In-Sb-Sn,
In-Sn-Zn

The “Atlas of Solders Phase Diagrams”

- The “Atlas of Phase Diagrams for Lead-Free soldering” has been published based on the data in the COST531 database



Bi-Zn system

Several authors worked on the theoretical description of the Bi-Zn system - e.g. Malakhov [00Mal], Cleari et al. [55Cle], Bale et al. [77Bal], Girard [85Gir] nebo Wang et al. [93Wan]. Kim and Sanders [03Kim] attempted to describe properly the segregation region in the liquid, using two interaction parameters LIZ only instead of six used by Malakhov. Unfortunately, their description is not consistent with unary parameters () of the „COST531“ database and therefore the data from Malakhov [00Mal] were used for this system. The changes of unary data were introduced and new value of $G(\text{HCP_ZN,BI,VA},0)$ from work of Moelans [03Moe] was accepted. New data were also introduced to model new experimental data from Vizdal et al. [07Viz] for solubility of Bi in HCP_ZN Zn ($G(\text{HCP_ZN,BI,ZN,VA},0)$).

References:

- [55Cle] Cleari, L., Florani, M., Valenti, V.: La Metallurgia Italiana, 1955, vol. 46, p. 773.
 [77Bal] Bale, C. W., Pelton, A. D., Rigaud, M.: Z. Metallkd., 1977, vol. 68, p. 69.
 [85Gir] Girard, C.: Ph.D. Thesis, Université de Provence, Marseille, 1985.
 [93Wan] Wang, Z. C., Yu, S. K., Sommer, F.: J. Chimie Phys. Physico-Chimie Biol., 1993, vol. 90, p. 379.
 [00Mal] Malakhov, D. V.: CALPHAD, 2000, vol. 24, p. 1.
 [03Kim] Kim, S. S., Sanders, T. H. Jr.: Z. Metallkd., 2003, vol. 94, p. 390.
 [03Moe] Moelans, N., Kumar, K. C. H., Wollants, P.: J. All. and Comp., 2003, vol. 360, p. 98.
 [07Viz] Vizdal, J., Braga, M. H., Kroupa, A., Richter, K. W., Soares, D., Malheiros, L. F., Ferreira, J.: CALPHAD - sent for publication, 2007

Table of invariant reactions

T / °C	Phases			Compositions / %		
254.5	RHOMBO_A7	LIQUID	HCP_ZN	0.016	0.081	0.998
271.4	LIQUID	RHOMBO_A7				
418.3	LIQUID 1	LIQUID 2	HCP_ZN	0.374	0.992	0.996
419.5	HCP_ZN	LIQUID				

Phase information

Phase Name	Common Name	Structure Type	Pearson Symbol
LIQUID	Liquid		
RHOMBO_A7	(Bi)	A7	hR2
HCP_ZN	(Zn)	A3 mod	hP2

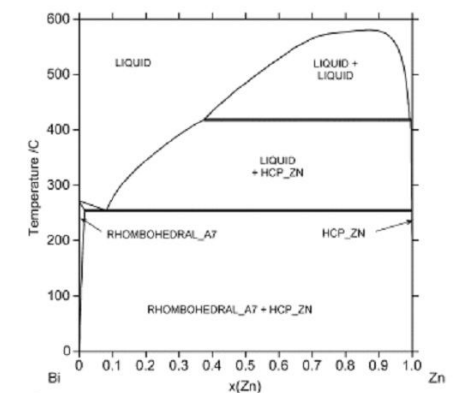


Fig. 1: ...

COST MP0602 - Scope

- Material under consideration
 - the high temperature solder
 - the substrate or contact material e.g. Cu, Ni-P, Pd
 - any legacy solder joint e.g. Pb-Sn
- Key solder systems identified
 - Zn-Al with Mg, Ge, Ga, Bi, Sn
 - Sn-Sb with Bi, Ag, Cu
 - Ag-Bi with Au, Sn
 - Au-Sn with Ag, Ge, Si
 - Zn-Sn with Bi, Ag, Cu

Ag, Al, Au, Bi, Co, Cu, Ga, Ge, In, Mg, Ni, P, Pb, Pd, Sb, Si, Sn, Ti, Zn

Final scope of the database

- the following set of systems for study

Ag-Cu-Ni-P-Pb-Pd-**Sn**

Al-Cu-Ga-Ge-Mg-Ni-P-Pb-Pd-Sn-**Zn**

Bi-**Co**-Cu-Ni-P-Pb-Pd-Sn-**Ti**

Ag-Au-Cu-Ni-P-Pb-Pd-**Sb-Sn**-Zn

Ag-Bi-Cu-Ga-Ni-P-Pb-Pd-Sn

Cu-Ni-P-Pb-Pd-**Sn-Zn**

- Please note

Contains new elements eg Al, Co, Ga, Ge, Mg, P, Ti

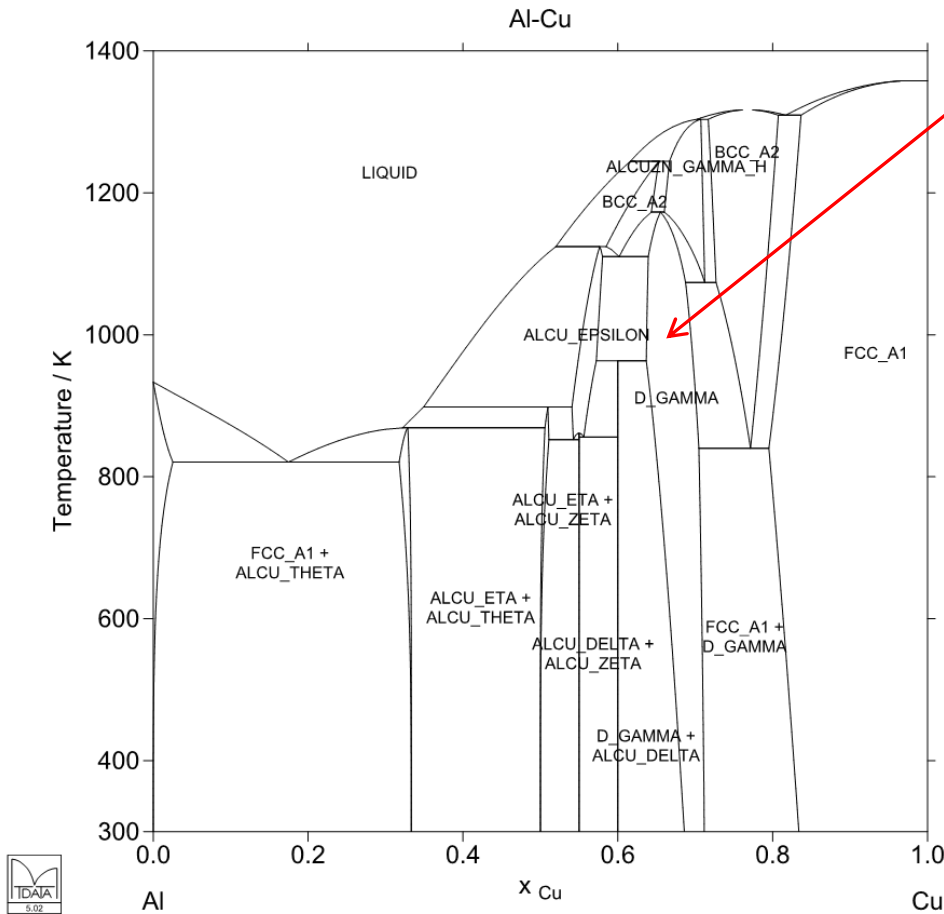
.... *but* leaves out In and systems such as Au-Bi, Bi-Sb and Bi-Zn

Modelling issues

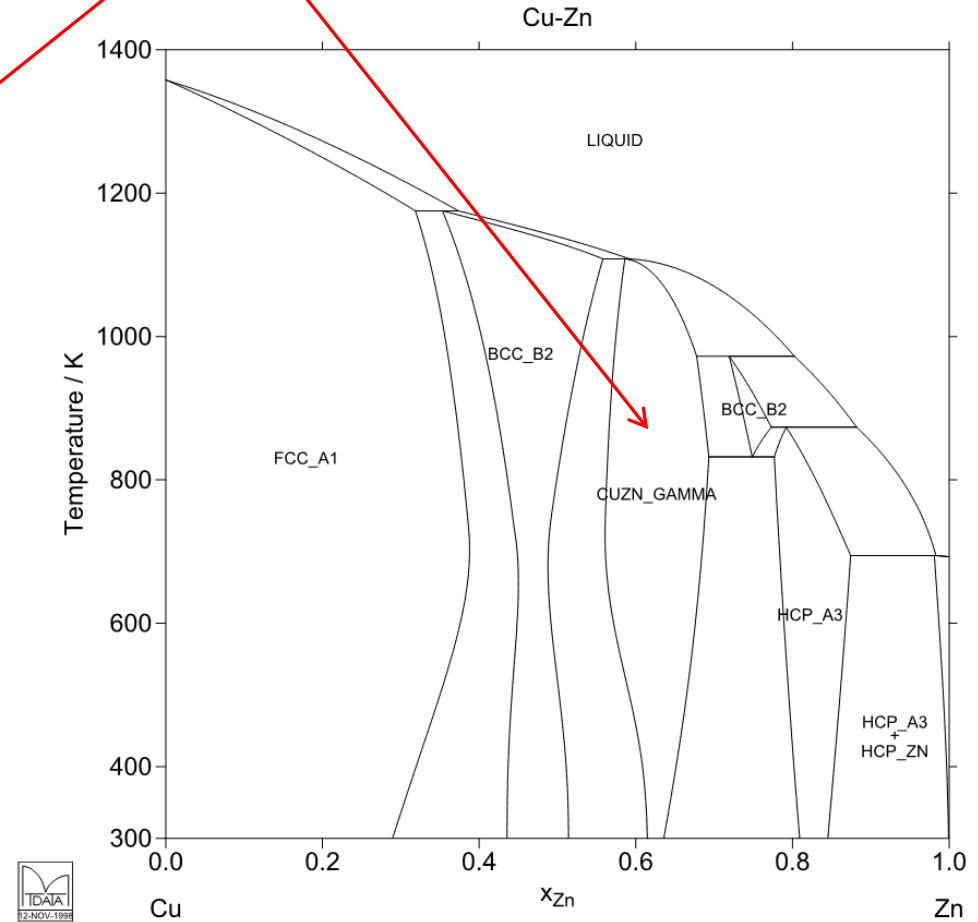
Database Construction: The need for internal consistency

- Models
 - Gamma brass
 - Cu-Zn
 - Cu_5Zn_8
 - $C/52$, $D8_2$
 - Al-Cu
 - Cu_9Al_4
 - $CP52$, $D8_3$

Database Construction: The need for internal consistency



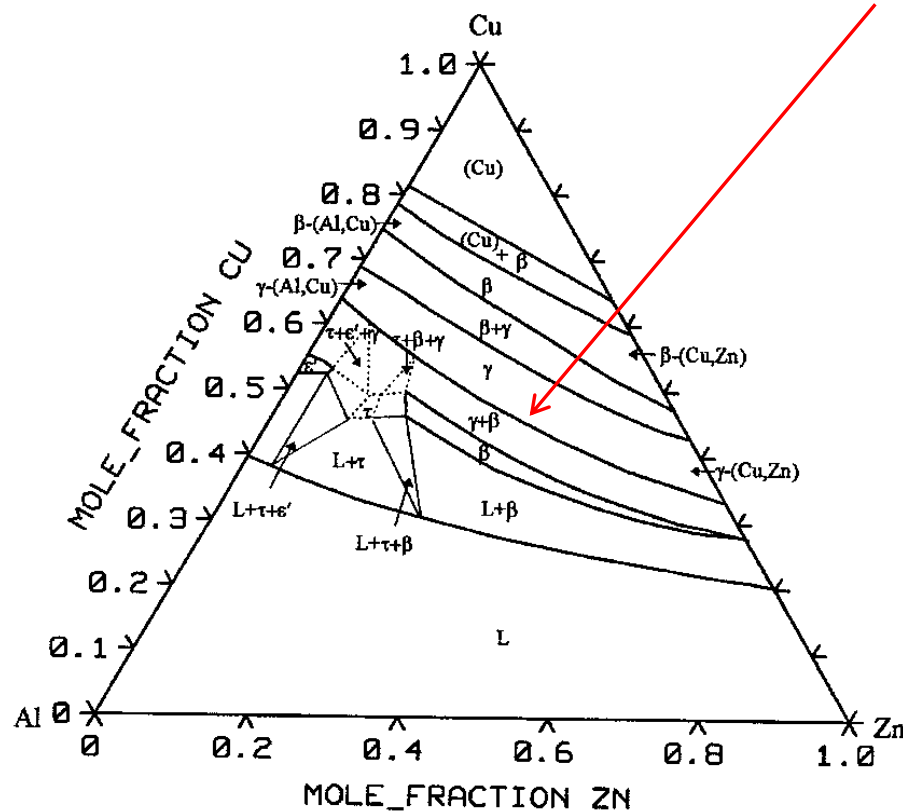
AL : AL CU : CU :



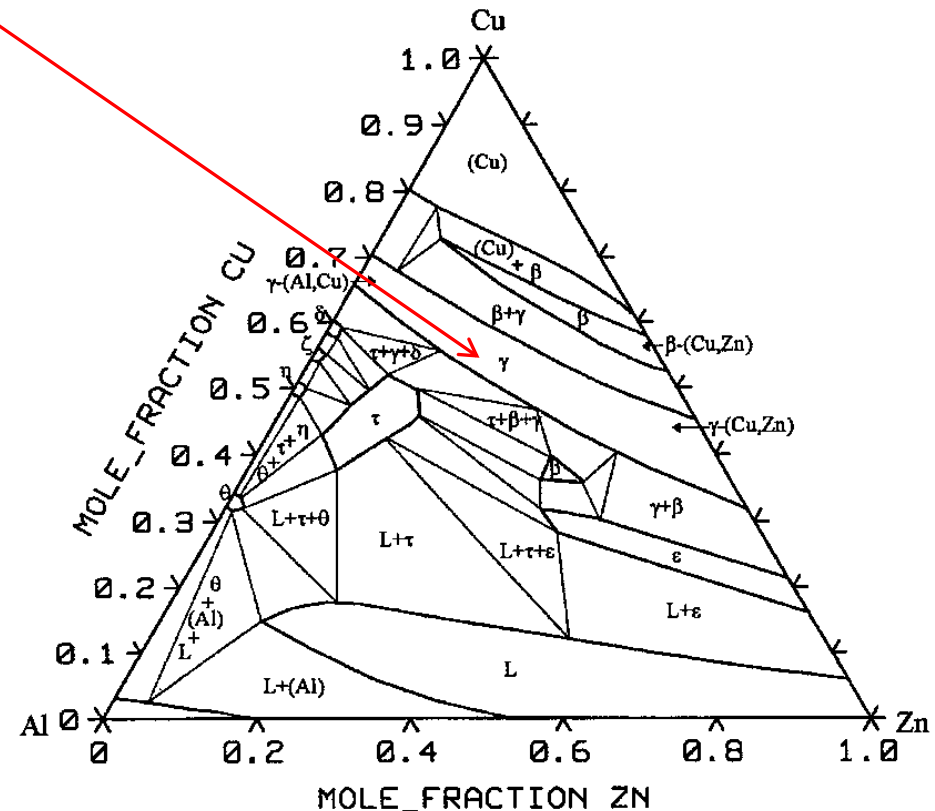
CU,ZN : CU,ZN : CU,ZN :CU,ZN :

Assessed Experimental Al-Cu-Zn Phase Diagram

Gamma brass

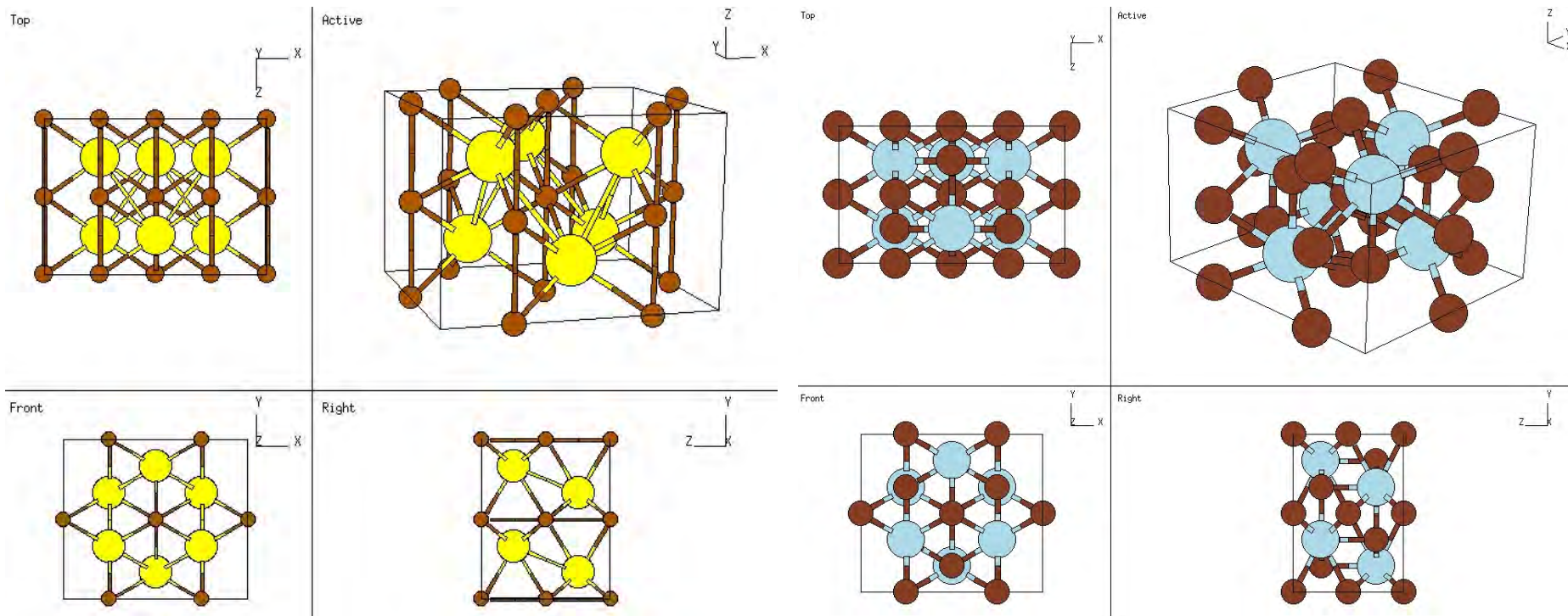


973K



773K

Database Construction: The need for internal consistency



NiAs prototype
Pearson symbol *hP4*

Ni₂In prototype
Pearson symbol *hP6*

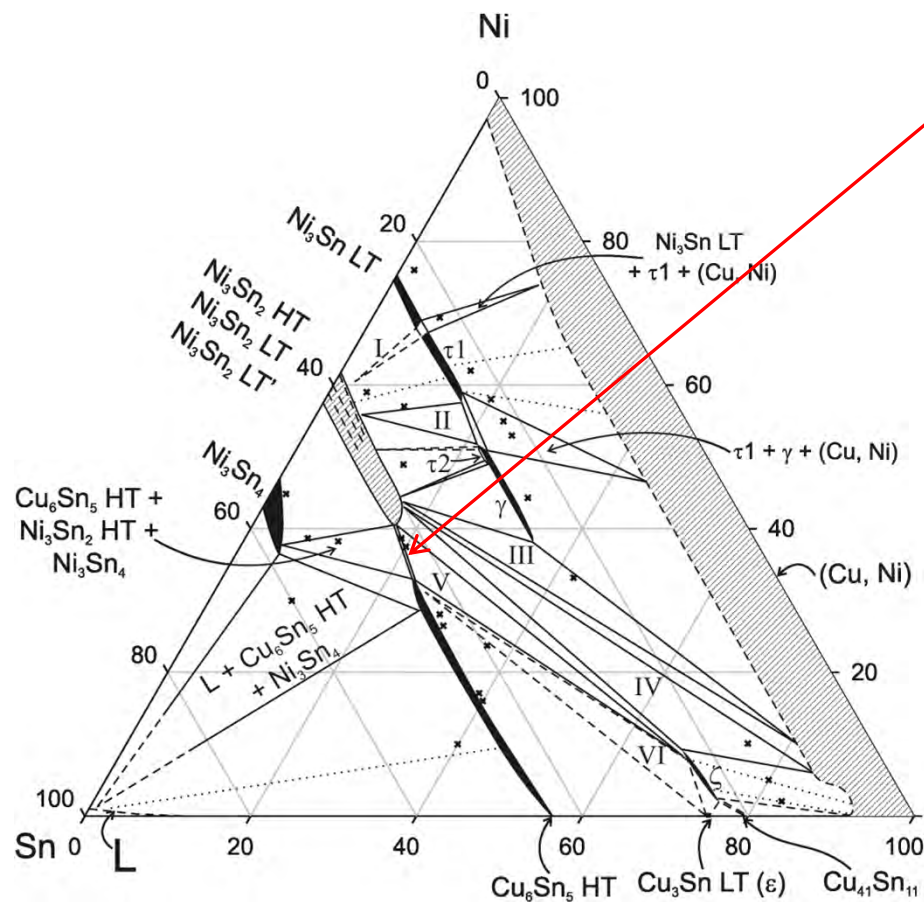
Thermodynamic models: NiAs and Ni₂In structures

- Cu₆Sn₅ phase has NiAs type structure (hP4)
particularly important, formed readily at the interface between Sn based solder and Cu substrate
- Ni₃Sn₂ has a structure based on the Ni₂In prototype (hP6)
Closely related to NiAs structure
- In Cu-Ni-Sn system
Ni dissolves appreciably in the Cu₆Sn₅ phase
Cu dissolves in Ni₃Sn₂
appears that the two phases do not form a continuous series of solid solutions
- Same situation in In-Ni-Sb system where there is a complete solubility between Ni₂In and NiSb

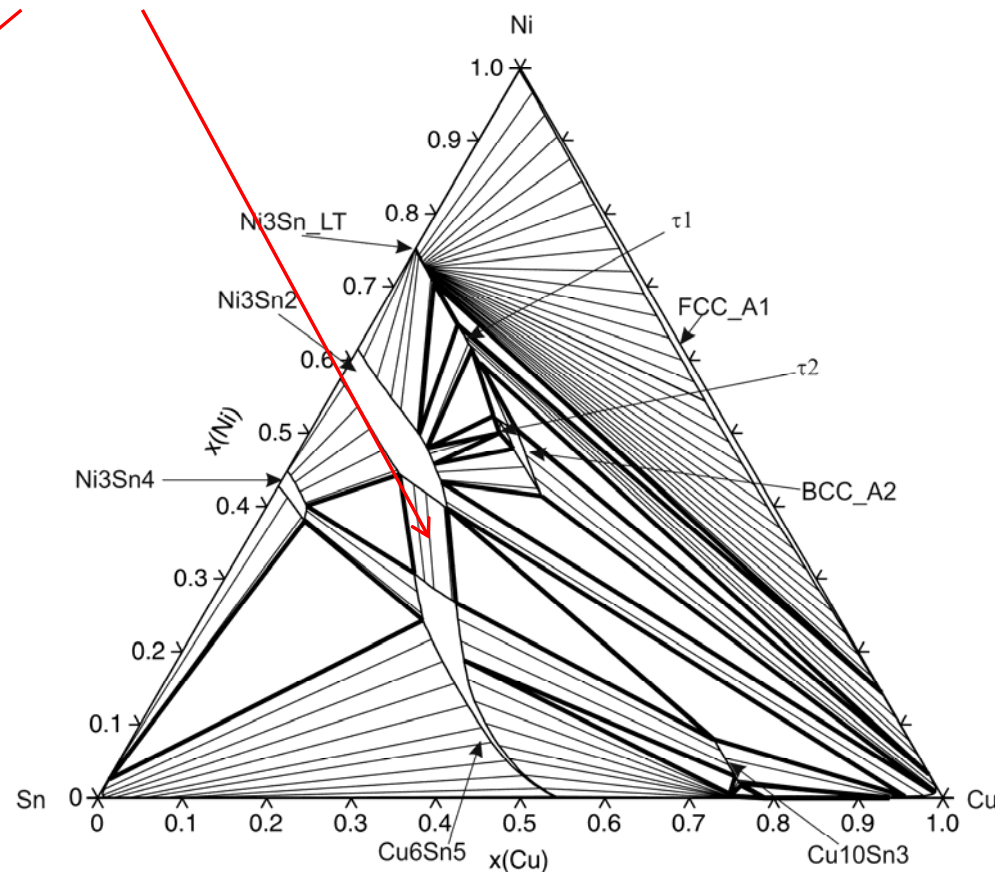


Database Construction: The need for internal consistency

Miscibility gap between Cu_6Sn_5 and Ni_3Sn_2

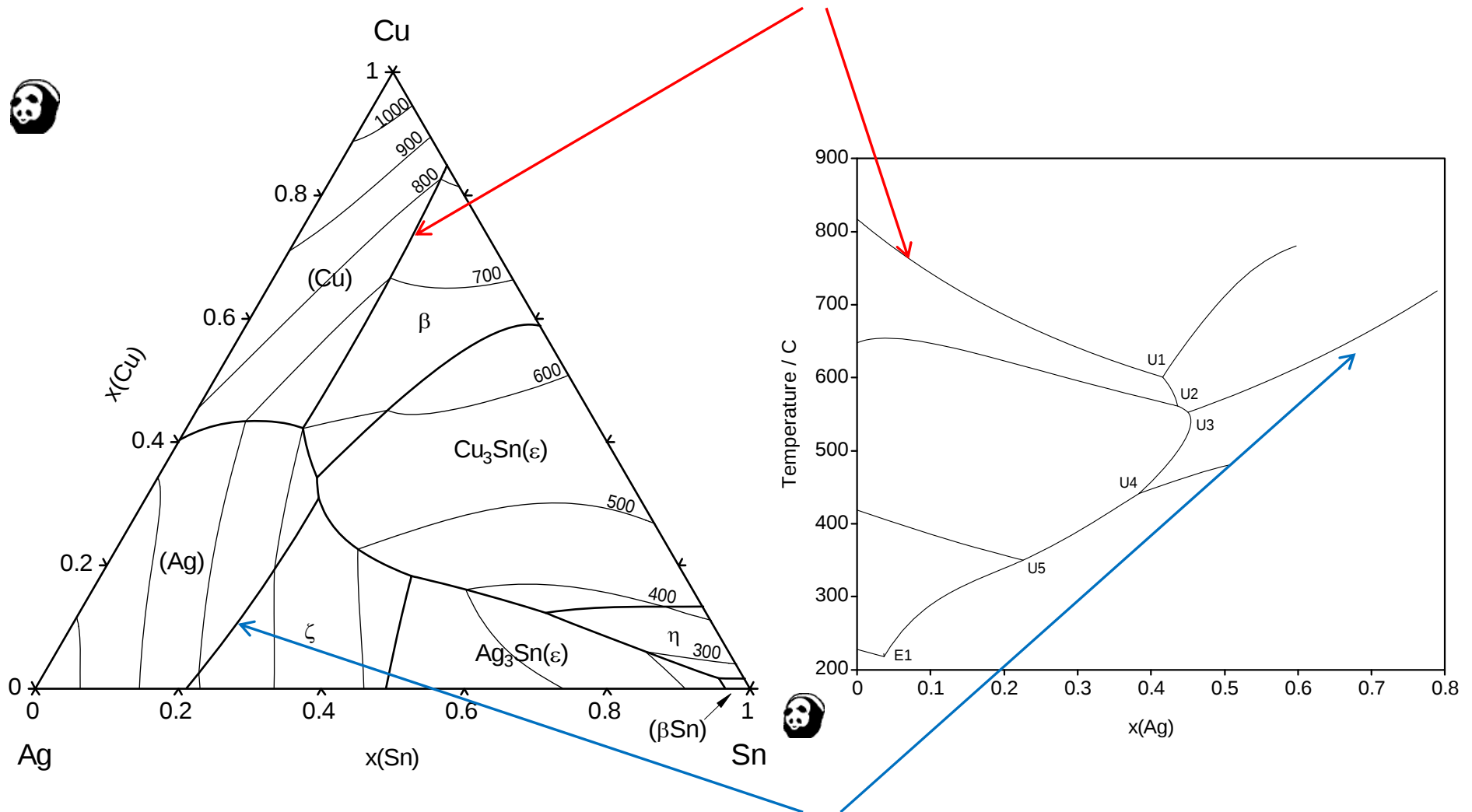


I = Ni_3Sn LT + τ_1 + Ni_3Sn_2 HT	IV = Ni_3Sn_2 HT + ζ + (Cu, Ni)
II = τ_1 + γ + Ni_3Sn_2 HT	V = Cu_6Sn_5 HT + Ni_3Sn_2 + ζ
III = Ni_3Sn_2 HT + γ + (Cu, Ni)	VI = Cu_6Sn_5 HT + ζ + ϵ



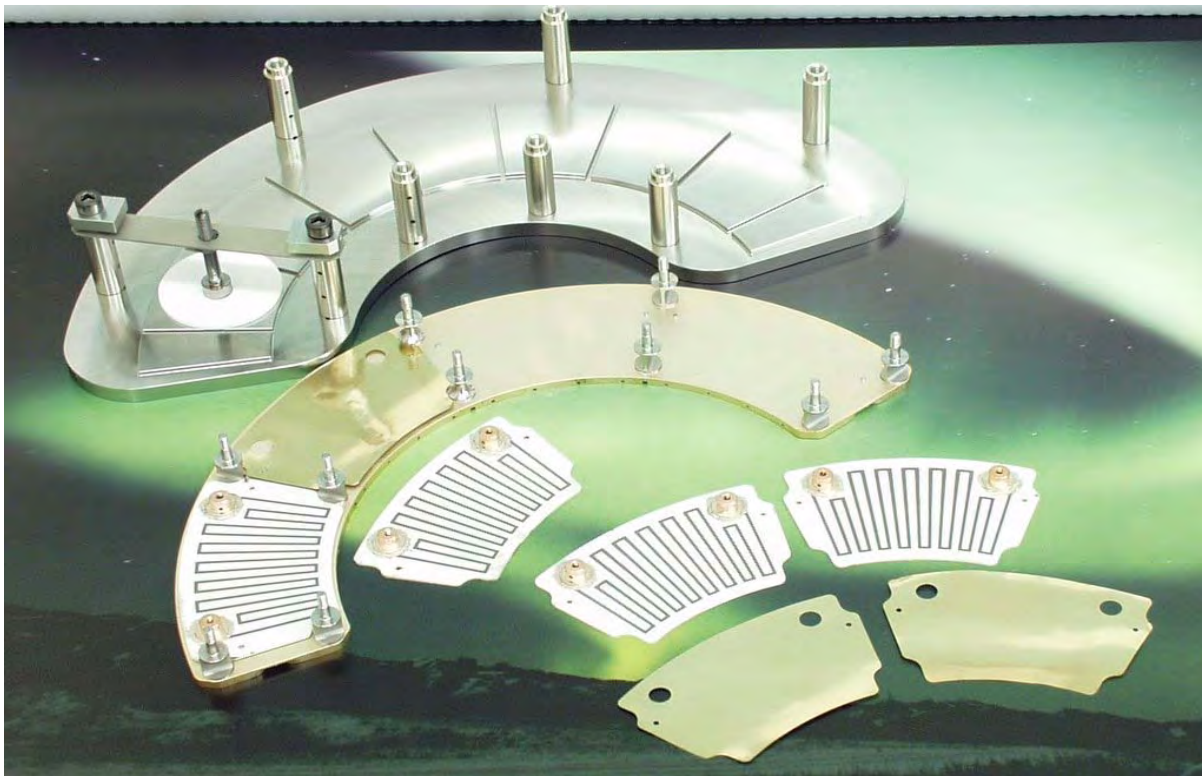
Applications

Liquidus projection for solders



Application - Au-Ge

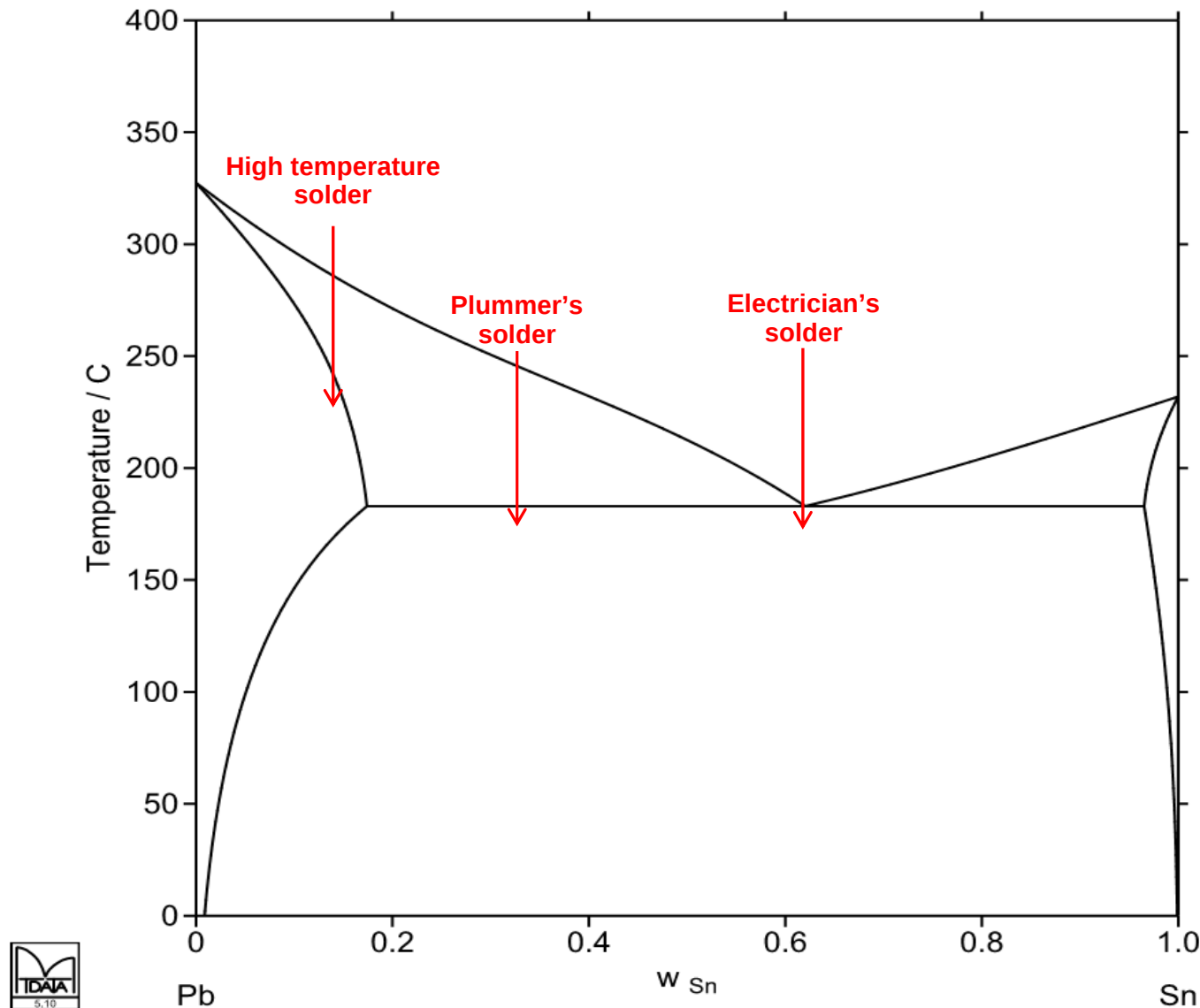
Heated conversion surface
assemblies for ESA / JAXA
Mission BepiColombo to Mercury
(Launch 2014)



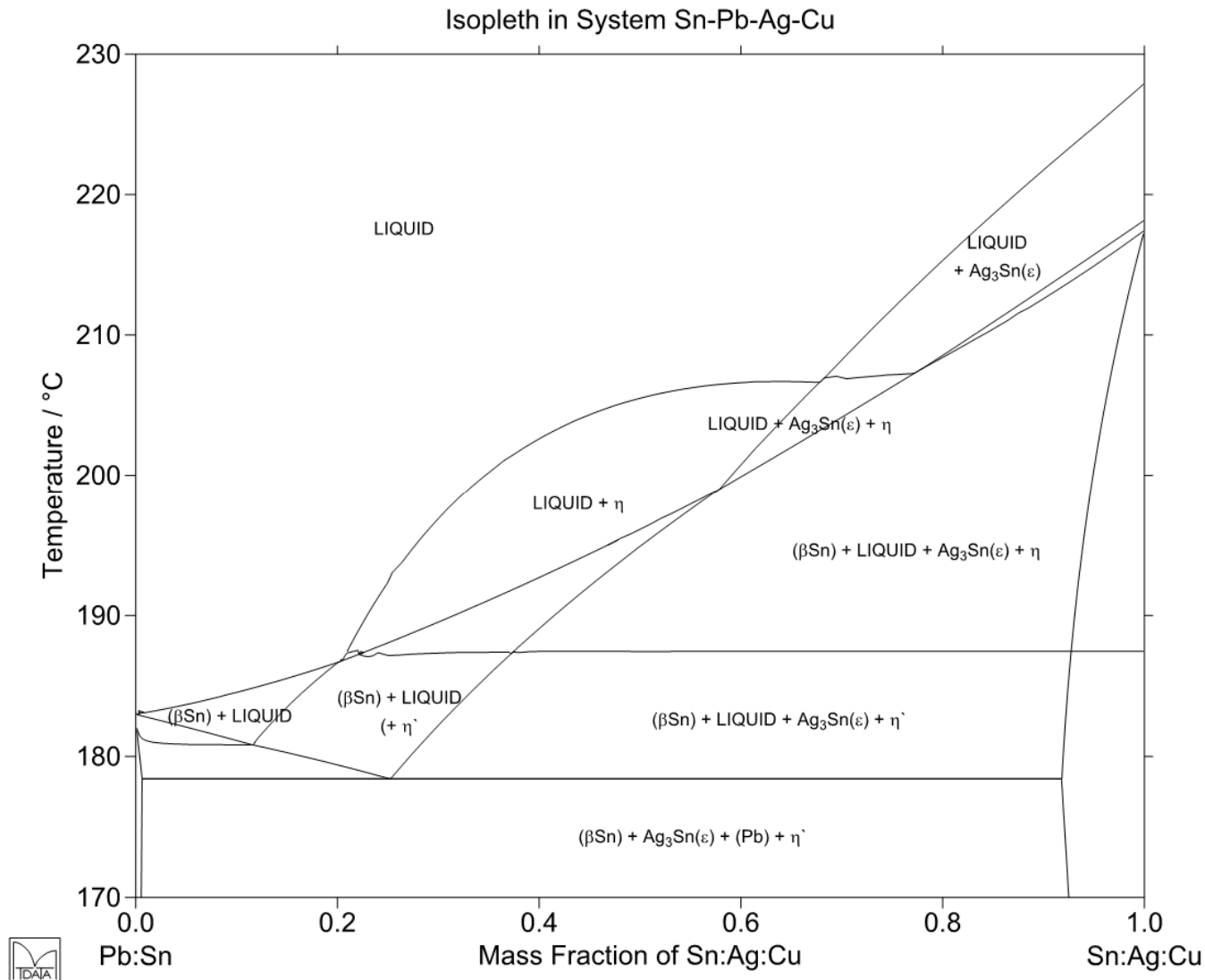
*Christian Leinenbach,
EMPA, Switzerland*

Investigating the reworking old solder joints

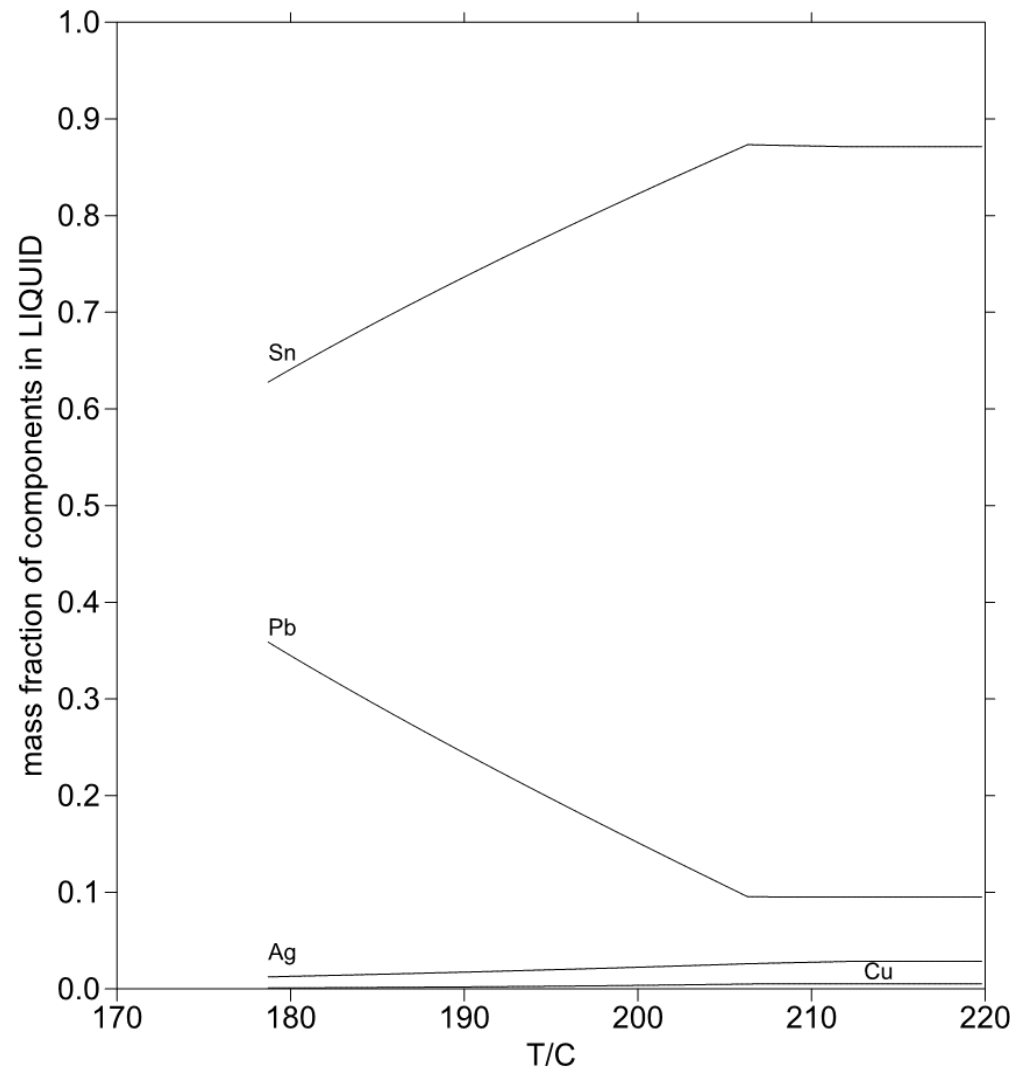
Phase Diagram for the Lead-Tin system



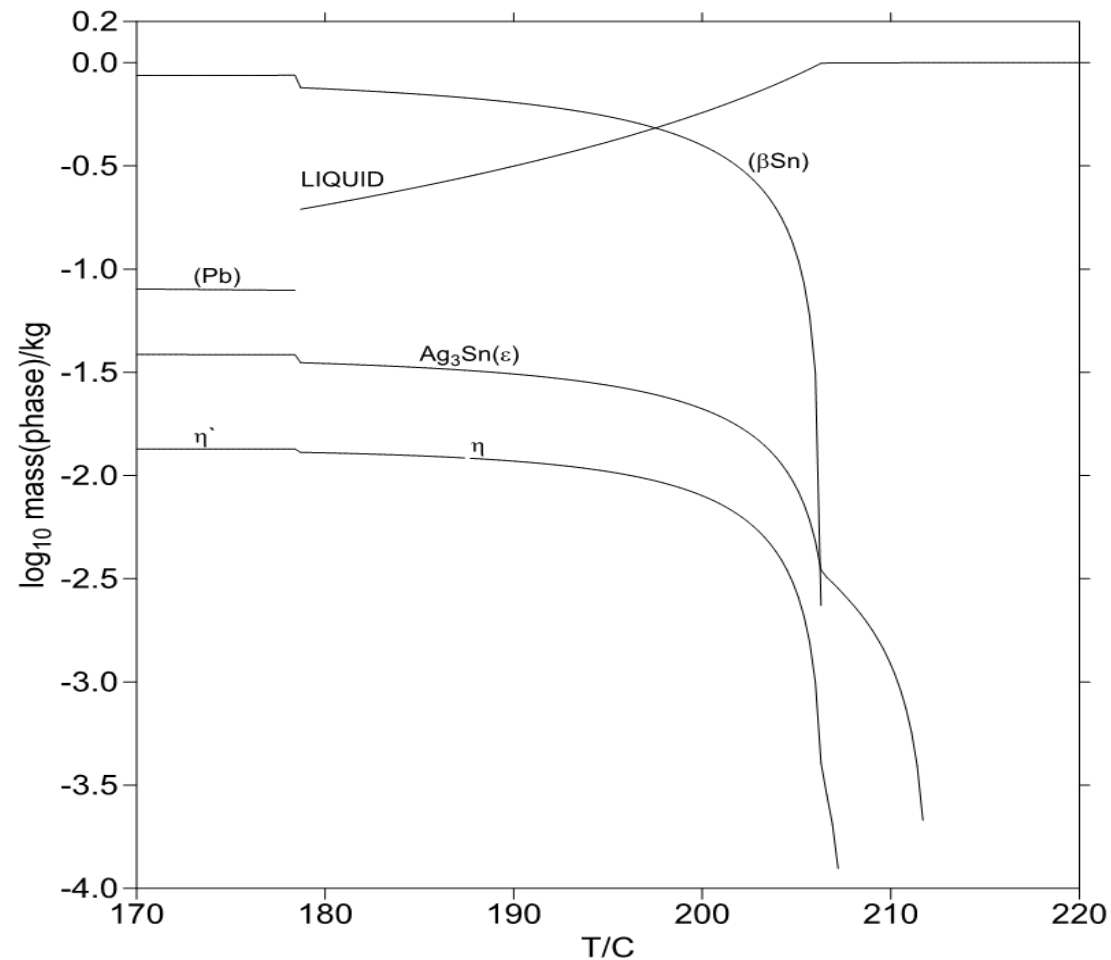
Repair: Mixing electrician's solder with lead free solder



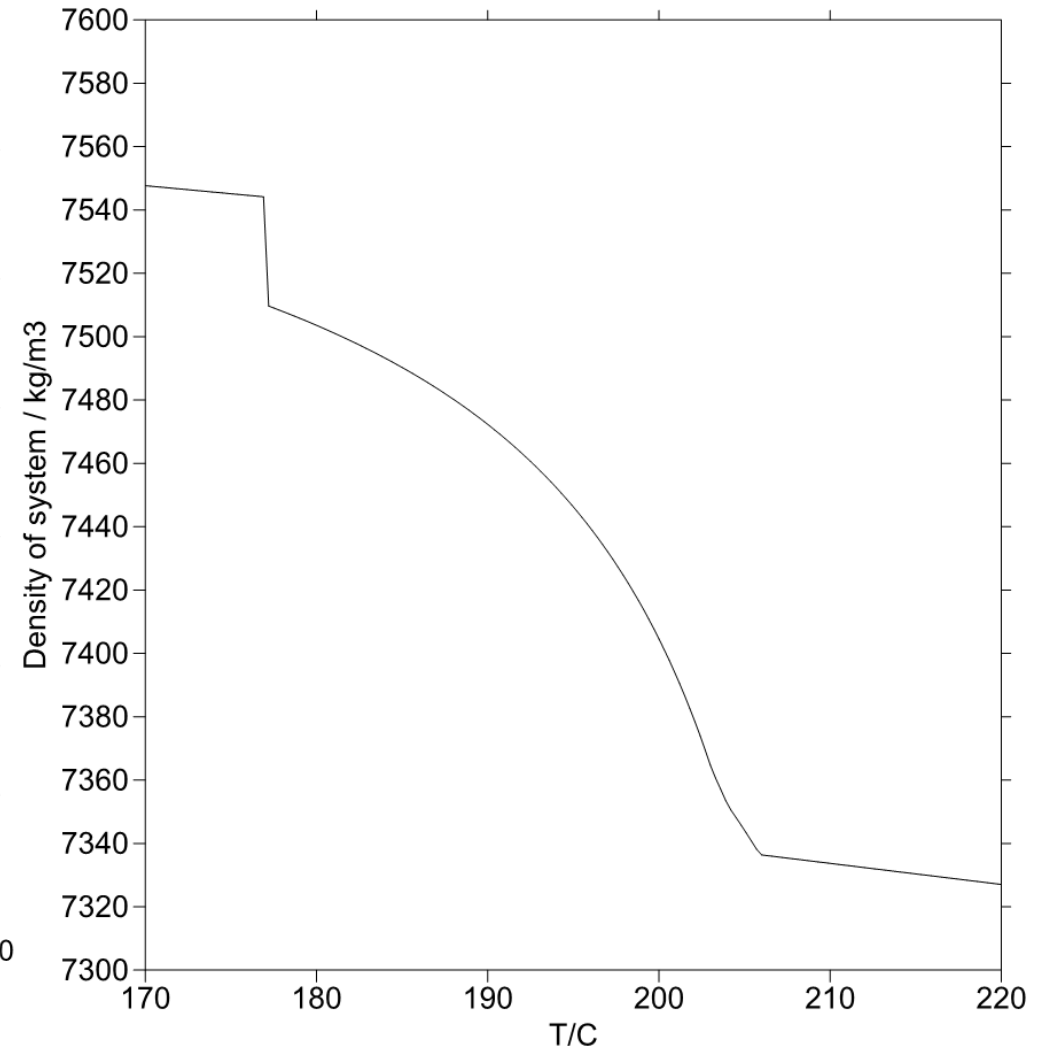
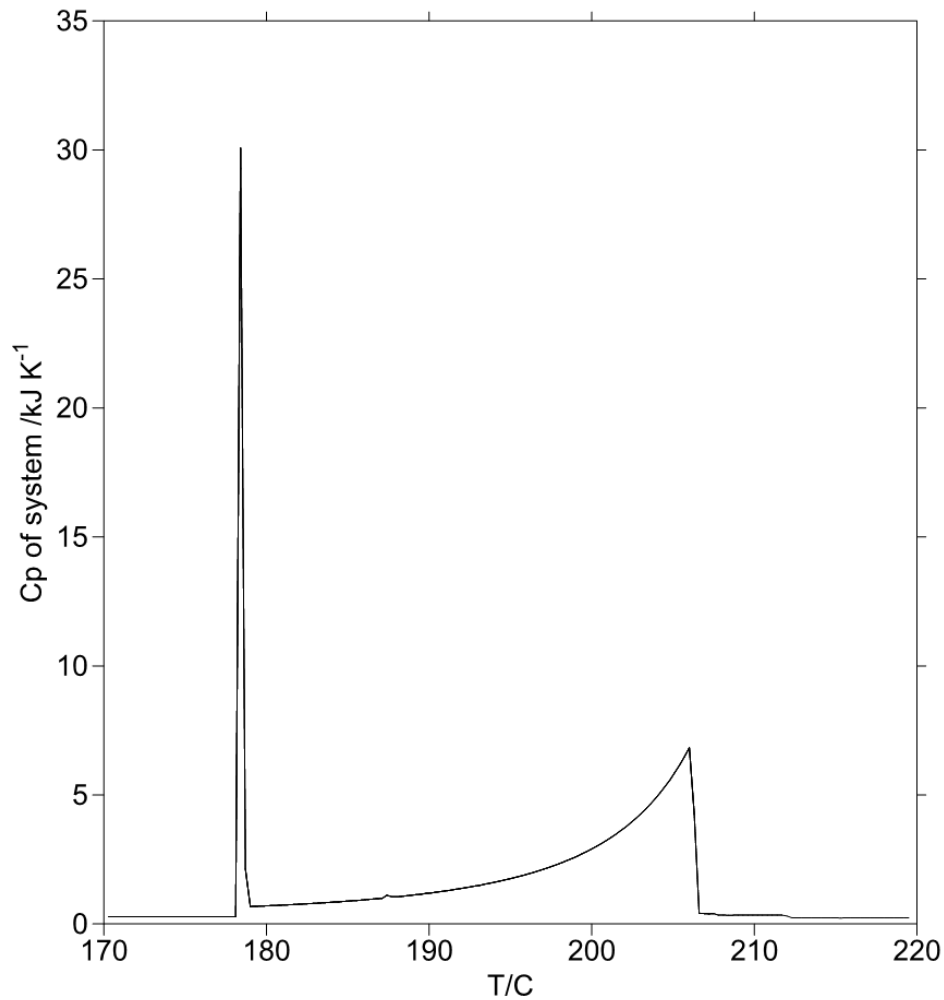
Liquid phase composition: cooling of a mixture of a lead-free solder and a Pb-Sn “electrician’s solder”



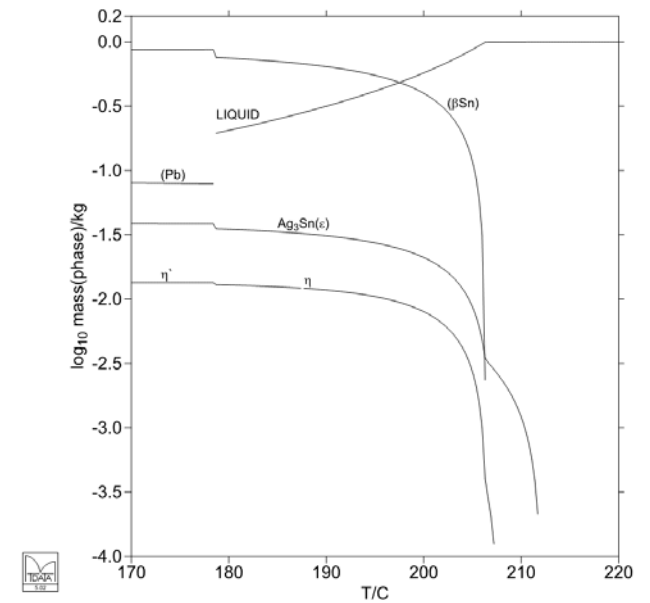
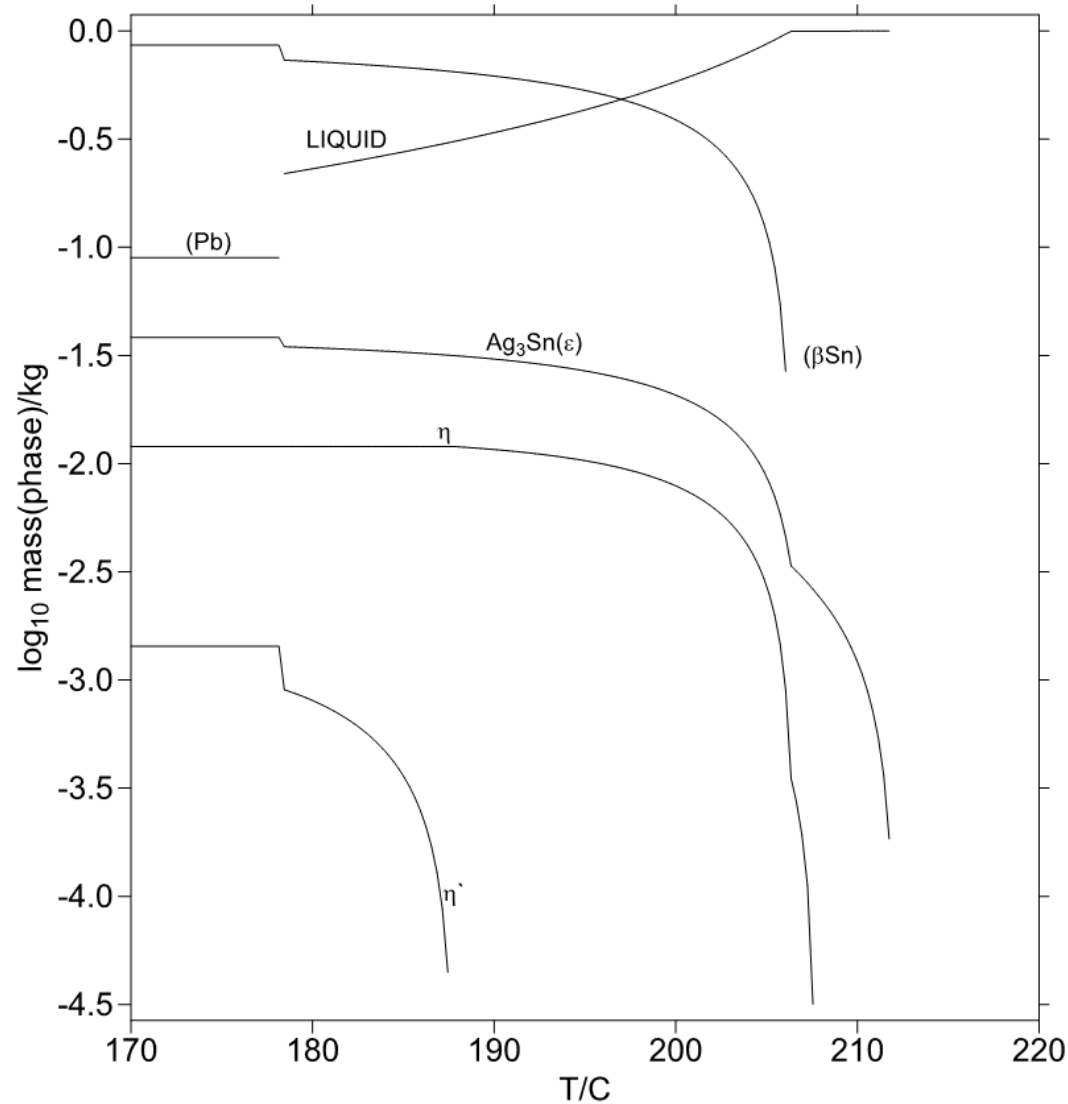
Plot of mass of phases with variation of temperature



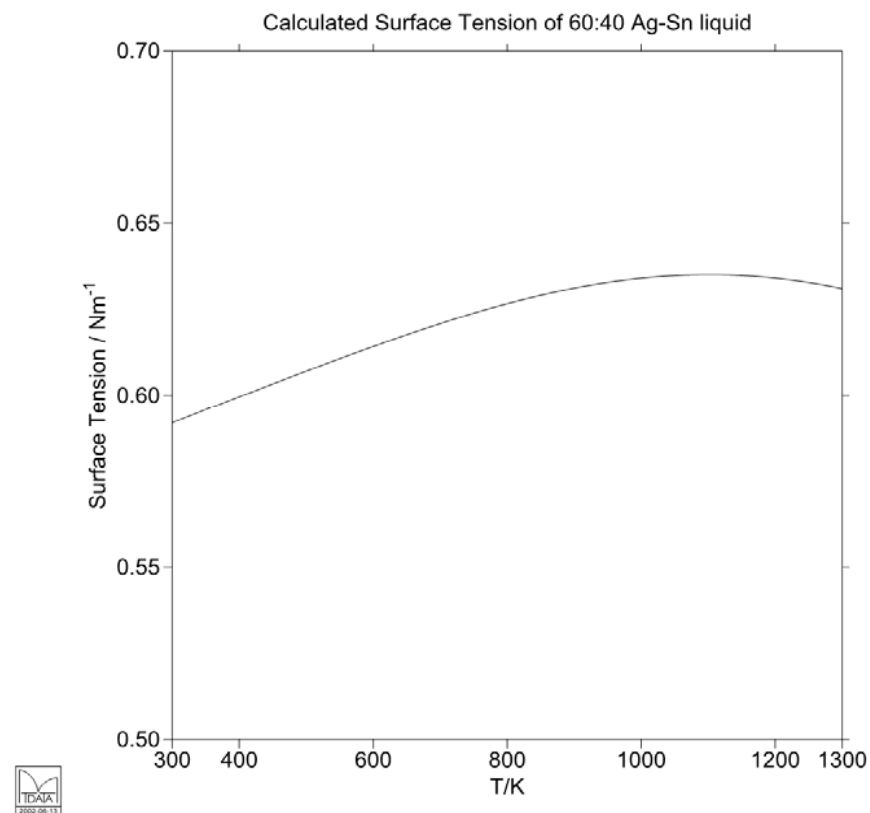
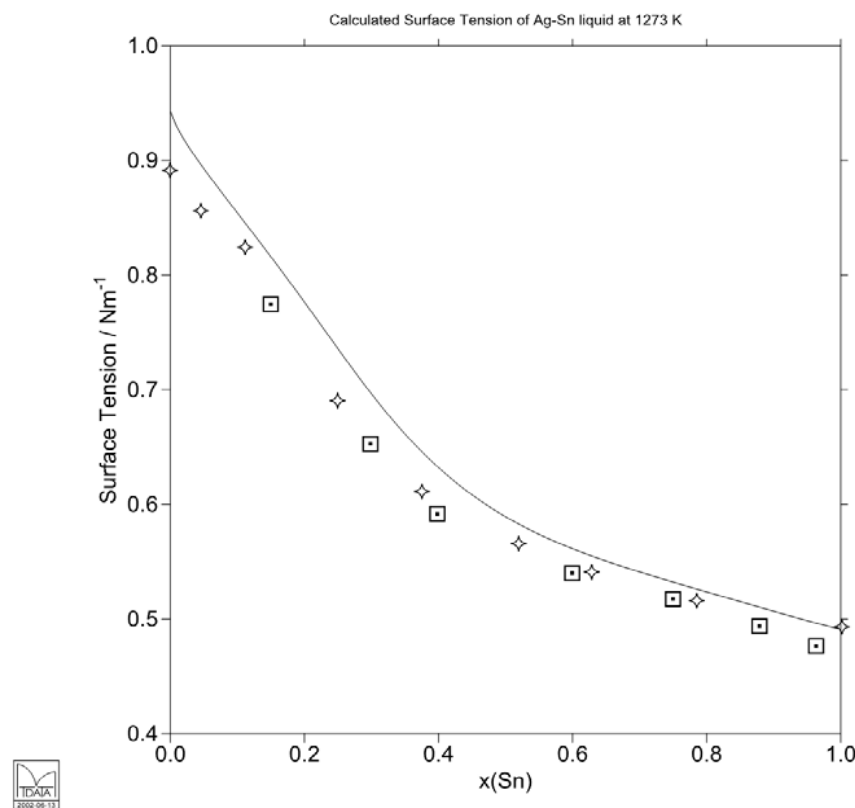
Calculated heat capacity and volume change



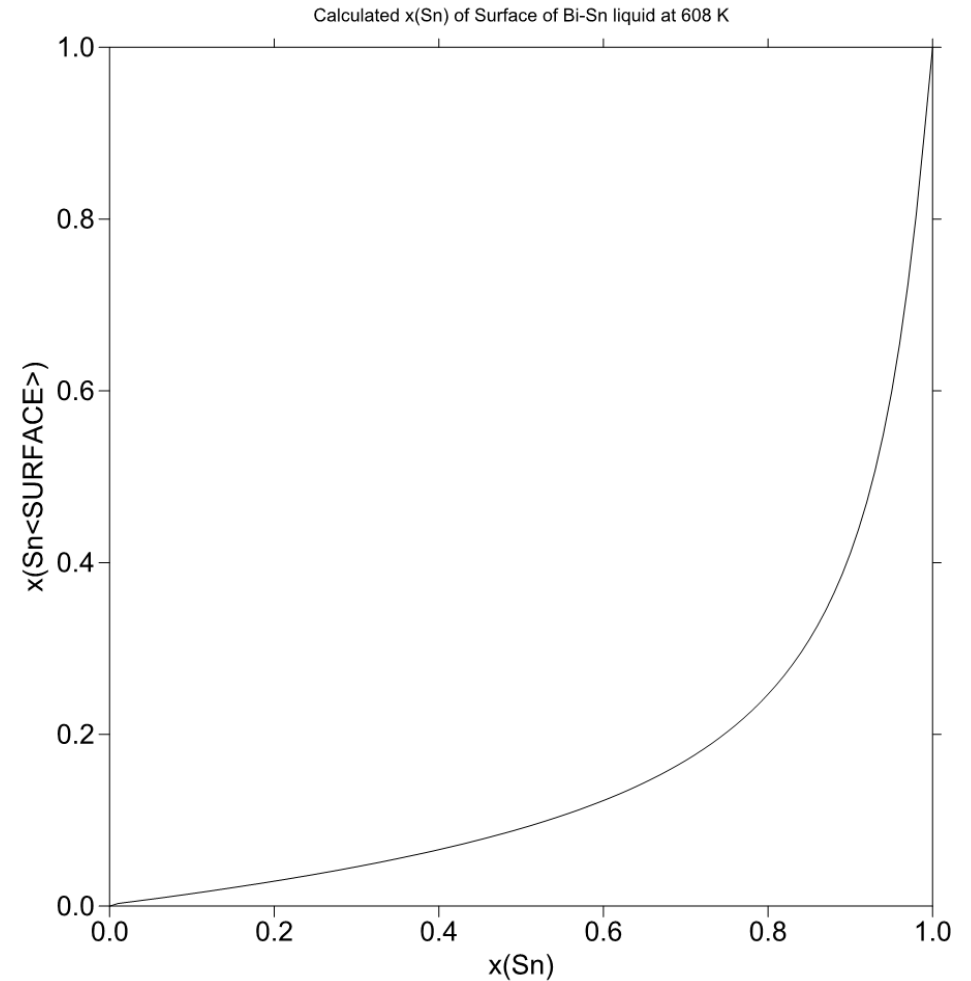
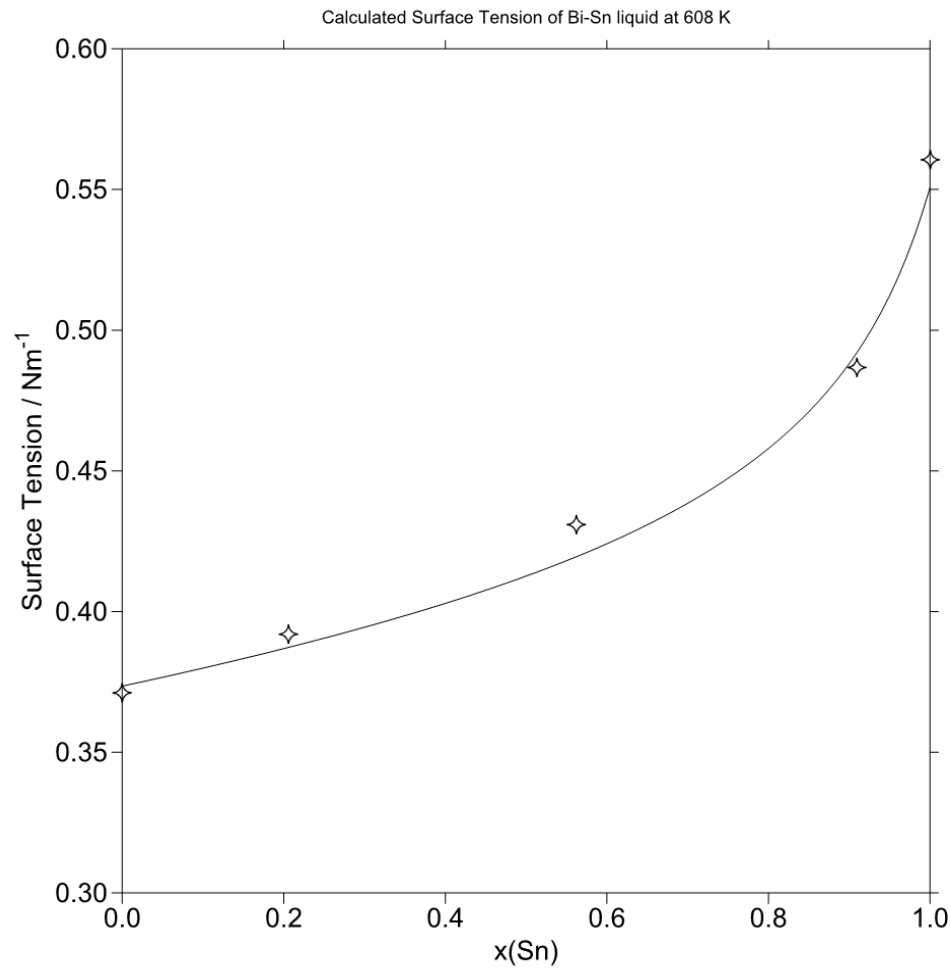
Scheil model of solidification



Other properties



Calculated surface tension and surface composition of Bi-Sn liquid



Final remarks

- Brief overview of European projects of solders
- What we have been trying to achieve
- Development and scope of thermodynamic databases
- Modelling issues
- Applications

Acknowledgements

- Aleš Kroupa (Institute of Physics of Materials, AS CR, Brno, Czech Republic)
- Andy Watson, (University of Leeds, UK.)
- Adéla Zemanová (Institute of Physics of Materials, AS CR, Brno, Czech Republic)
- Jan Vřešťál, Pavel Brož (Institute of Theor.and Phys. Chemistry, Masaryk University, Brno, Czech Republic)

Thanks !